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Preliminary communication

Carbonyl olefination with α -stannyl ester enolates: a new synthesis of α , β -unsaturated esters

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Abstract

The reaction of α -stannyl ester enolates with carbonyl compounds is described. α , β -Unsaturated esters are obtained in good yields. A reaction mechanism is proposed.

 α -Silyl ester enolates react with aldehydes and ketones to give α,β -unsaturated esters [1]. This reaction has become an important method for the synthesis of this type of organic compounds [2]. In this communication we report that analogous carbonyl olefination are possible with α -stannyl esters [3], providing a new application of these organotin compounds in synthetic organic chemistry.

Addition of tert-butyl (tributylstannyl)acetate, 1 [3], to tetrahydrofuran (THF) solutions of lithium diisopropylamide (LDA) at -78 °C produces the lithium ester

$$Bu_{3}SnCH_{2}CO_{2}'Bu \xrightarrow{LDA, THF} \left[Bu_{3}Sn\bar{C}HCO_{2}'Bu\right] \xrightarrow{PhCHO}$$
(1)
(2)



Scheme 1.

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Ester "	Carbonyl compound	Product ^a	% Yield ${}^{b}(E:Z \text{ ratio}) {}^{c}$
Ph ₃ SnCH ₂ CO ₂ ^t Bu	benzaldehyde	PhCH=CHCO ₂ ^t Bu	82(>95:<5)
	cyclohexanone	CHCO ₂ 'Bu	73
	2-phenlypropanal	$Ph(Me)CH - CH = CHCO_2^{t}Bu$	86(>95:<5)
Ph ₃ SnCH ₂ CO ₂ Me	benzaldehyde	$PhCH = CHCO_2Me$	68(>95:<5)
	cyclohexanone	CHCO ₂ Me	78
	2-phenylpropanal	$Ph(Me)CH - CH = CHCO_2Me$	69(> 95 : < 5)
Ph ₃ SnCHCO ₂ Me	benzaldehyde	$PhCH = C(Me)CO_2Me$	78(90:10)
Me		_	
	cyclohexanone	$\langle \rangle = C(Me)CO_2Me$	79
	2-phenylpropanal	$Ph(Me)CH - CH = C(Me)CO_{2}Me$	68(2:8)
Bu ₃ SnCH ₂ CO ₂ ^t Bu	benzaldehyde	PhCH=CHCO ¹ Bu	60(> 95 : < 5)

Table 1			
Reaction of α -stannyl	esters with	carbonyl	compounds

^{*a*} All new compounds have spectral and analytical data in agreement with the proposed structure. ^{*b*} Isolated yield after purification by flash chromatography on silica gel. ^{*c*} Ratio of Isomers estimated by HPLC

enolate 2 (Scheme 1). Subsequent treatment with one equivalent of benzaldehyde gave a mixture of two major products: the expected product tert-butyl cinnamate, 5, in 60% yield, and tert-butyl bis(α -hydroxybenzyl)acetate, 6, in 12% yield [4*]. By inverse addition to five equivalents of benzaldehyde, the yield of 6 was increased to 22%.

Conceivably, compound 6 could be formed by reaction of benzaldehyde with β -stannyloxy enolate, 4, which in turn was formed from β -oxido stannane, 3, by 1,3-migration of tin from carbon to oxygen [5*]. The isolation of 6 constitutes strong evidence in favor of 4 as a key intermediate in these reactions [6*].

We also examined this reaction with α -triphenylstannyl ester enolates. The starting esters were conveniently prepared by a modification of our published procedure [3]. Thus, triphenylstannylation of carboxylic esters with the aid of LDA afforded the corresponding C-stannylated products in high yields.

Reaction of α -triphenylstannyl ester enolates with carbonyl compounds gave the desired α,β -unsaturated esters. The results are shown in Table 1. As can be noted, the yields are good and the reaction proceeds with both aldehydes and ketones. High *E* selectivity is observed which is consistent with the proposed mechanism (see Scheme 1). Elimination of the β -stannyloxy group in intermediate 4 should occur from the conformation having less steric interactions.

 α -Silyl esters have been used for the synthesis of α,β -unsaturated esters for many years [1,2]. However, access to α -silyl esters by direct C-silylation of ester enolates is possible only with tert-butyl acetates [7] or by the use of the rather expensive reagent diphenylmethylchlorosilane [8]. In contrast, a wide variety of

^{*} Reference number with asterisk indicates a note in the list of references.

 α -stannyl esters are readily available in high yields by C-stannylation of lithium ester enolates with either tributyl- or triphenyltin chloride [3]. Therefore, this new synthesis of α,β -unsaturated esters, from carbonyl compounds and α -stannyl esters, has distinct advantages. The results described here offer both a convenient synthetic alternative and an incentive to explore new uses of α -stannyl esters in organometallic chemistry.

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- 4 Besides having spectral and analytical data in accord with the proposed structure, the identity of compound **6** was confirmed by mixed melting point determination with an independently synthesized sample.
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